

## Total Synthesis of ( $\pm$ )-(*E*)-8 $\beta$ ,17-Epoxyabd-12-ene-15,16-dial

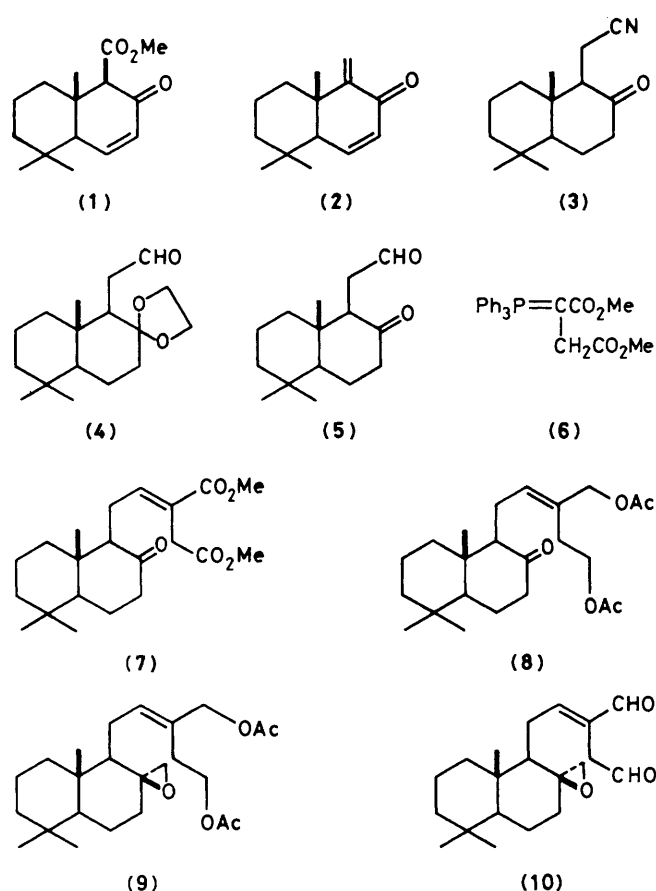
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The title labdanoid dialdehyde (**10**) which has high antifungal activity was synthesized stereoselectively from the *trans*-decalone derivative (**1**).

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In connection with our previous synthesis<sup>1</sup> of polygodial and warburganal which possess a wide spectrum of biological activity including insect antifeeding, antimicrobial, cytotoxic, plant growth regulation, molluscicidal, and anticomple-



mental properties, we attempted the synthesis of the labdanoid dialdehyde (10) which was isolated from the seeds of *Afromomum daniellii* in 1979<sup>2</sup> and has the same functional groups as polygodial. In this paper, we describe the first synthesis of the labdanoid dialdehyde (10) from the *trans*-decalone derivative (1).<sup>3</sup>

The cyano-ketone (3), m.p. 80–82 °C [ $\nu_{\max}(\text{CCl}_4)$  2250 and 1715  $\text{cm}^{-1}$ ], was synthesized from the *exo*-methylene enone (2)<sup>4</sup> by the Michael addition of a C<sub>1</sub> unit at C-11 (KCN, NH<sub>4</sub>Cl, dimethylformamide–H<sub>2</sub>O, room temp.) followed by hydrogenation (Pd/C, EtOH) [90% yield from (2)]. Compound (3) was protected (ethylene glycol, *p*-MeC<sub>6</sub>H<sub>4</sub>SO<sub>3</sub>H), reduced (di-isobutylaluminium hydride, hexane), and deprotected with acid (50% acetic acid) to give the keto-aldehyde (5) [quantitative yield from (3)], [(5) (oil):  $\nu_{\max}(\text{CCl}_4)$  1720 and 1710  $\text{cm}^{-1}$ ; <sup>1</sup>H n.m.r. (CCl<sub>4</sub>)  $\delta$  0.73 (3H, s), 0.88 (3H, s), 1.00

(3H, s), and 9.73 (1H, s)]. Next, (5) was treated with the phosphorane (6)<sup>5</sup> to give the keto-diacetate (7) having (*E*)-stereochemistry<sup>6</sup> (72.4%, benzene reflux temp., 20 h) [(7) (oil):  $\nu_{\max}(\text{CCl}_4)$  1740 and 1710  $\text{cm}^{-1}$ ; <sup>1</sup>H n.m.r. (CCl<sub>4</sub>)  $\delta$  0.76 (3H, s), 0.89 (3H, s), 0.99 (3H, s), 3.34 (2H, s), 3.65 (3H, s), 3.70 (3H, s), and 6.69 (1H, t, *J* 6 Hz)]. However, the reaction of the acetal-aldehyde (4) with the phosphorane (6) gave the acetal of the keto-diacetate (7) in only 8% yield owing to the steric hindrance of the bulky ethylene acetal group. Compound (7) was converted into the keto-diacetate (8) [ $\nu_{\max}(\text{CCl}_4)$  1740 and 1710  $\text{cm}^{-1}$ ; <sup>1</sup>H n.m.r. (CCl<sub>4</sub>)  $\delta$  0.76 (3H, s), 0.88 (3H, s), 1.00 (3H, s), 2.02 (6H, s), 4.08 (2H, t, *J* 6 Hz), 4.40 (2H, s), 5.42 (1H, t, *J* 7 Hz)] by the usual method [i, ethylene glycol, *p*-MeC<sub>6</sub>H<sub>4</sub>SO<sub>3</sub>H; ii, LiAlH<sub>4</sub>, diethyl ether; iii, Ac<sub>2</sub>O, pyridine; iv, 50% AcOH; 65.2% yield from (7)]. Treatment of (8) with dimethylsulphonium methylide followed by acetylation gave the oxirane (9) (78% conversion yield), [(9) (oil):  $\nu_{\max}(\text{CCl}_4)$  1740, 1230, and 1025  $\text{cm}^{-1}$ ; <sup>1</sup>H n.m.r. (CCl<sub>4</sub>)  $\delta$  0.92 (9H, s), 2.00 (3H, s), 2.04 (3H, s), 2.18 (1H, d, *J* 3.5 Hz), 2.43 (1H, d, *J* 3.5 Hz), 4.04 (2H, t, *J* 6 Hz), 4.42 (2H, s), and 5.36 (1H, br. m)]. Finally, hydrolysis (K<sub>2</sub>CO<sub>3</sub>, aqueous MeOH, room temp.) of (9) afforded the corresponding diol, which without purification was oxidized (oxalyl chloride, Me<sub>2</sub>SO, CH<sub>2</sub>Cl<sub>2</sub>, NEt<sub>3</sub> or NCS, Me<sub>2</sub>S, AgBF<sub>4</sub>, NEt<sub>3</sub>) to the dialdehyde (10) [43% yield from (9)] [(10) (oil)  $\nu_{\max}(\text{CCl}_4)$  1730, 1690, and 1640  $\text{cm}^{-1}$ ; u.v. (EtOH)  $\lambda_{\max}$  234 nm; <sup>1</sup>H n.m.r. (CDCl<sub>3</sub>)  $\delta$  0.88 (3H, s), 0.91 (3H, s), 0.93 (3H, s), 2.30 (1H, d, *J* 3.5 Hz), 2.44 (1H, d, *J* 3.5 Hz), 3.40 (2H, br. s), 6.62 (1H, t, *J* 7 Hz), 9.32 (1H, s), 9.56 (1H, t, *J* 1 Hz)], which had identical <sup>1</sup>H n.m.r. and i.r. spectra with the natural product.

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